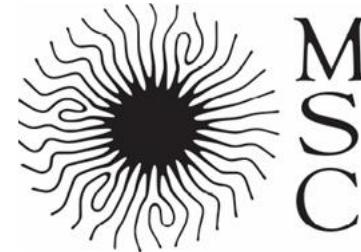


Effects of brake wear nanoparticles on respiratory cells

Chloé PUISNEY, Evdokia OIKONOMOU, Sophie NOWAK, Alexandre CHEVILLOT,
Jean-François BERRET and Armelle BAEZA-SQUIBAN

5th International NanoSafe Conference, Grenoble, 9th November 2016



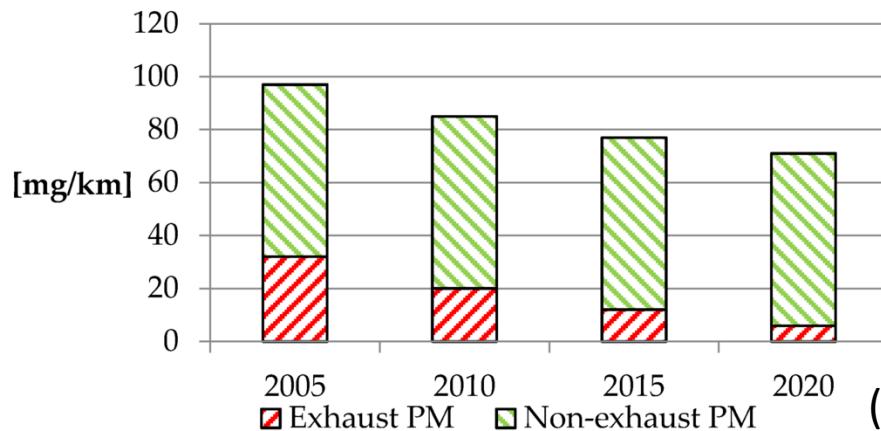
Non-exhaust

General considerations (1)

Exhaust



Average urban PM₁₀ emissions by passenger cars



(Hooftman *et al*, Energies, 2016)



- Brake wear contribution (Grigoratos *et al*, Environmental Science Pollution Research, 2015) :
 - up to 55 % by mass to total non-exhaust traffic-related PM₁₀
 - up to 21 % by mass to total traffic-related PM₁₀ emissions

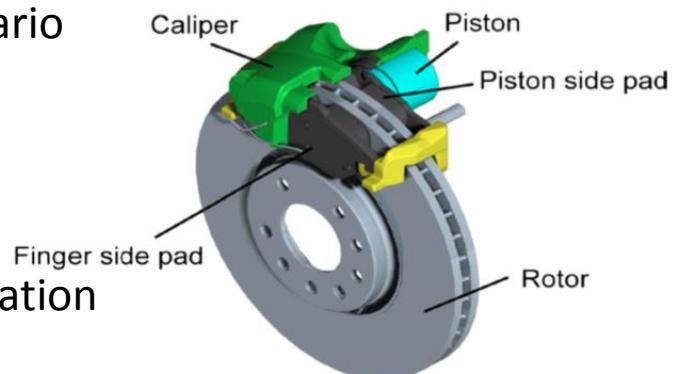
General considerations (2)

What is already known about brake wear particles ?

- Braking scenario associated with temperature ↑ + friction strength
 - Surface chemistry alteration
 - Material degradation
 - Favorable to particles generation
- Wide range of size : from few nanometers to several micrometers

(Thorpe and Harrisson, Science of Total Environment, 2008)

→ Direct link between particles size and braking scenario



(Wahlström *et al*, 2009)

Experimental strategy



Physicochemical characterization (1)

X-ray Fluorescence	
Chemical element	Composition (%)
Mg	1,6
Al	11,0
Si	13,1
S	6,9
K	2,1
Ca	3,5
Ti	1,3
Cr	1,8
Mn	0,2
Fe	29,1
Cu	16,7
Zn	7,7
Zr	1,0
Sn	1,1
Bi	1,6

Physicochemical characterization (1)

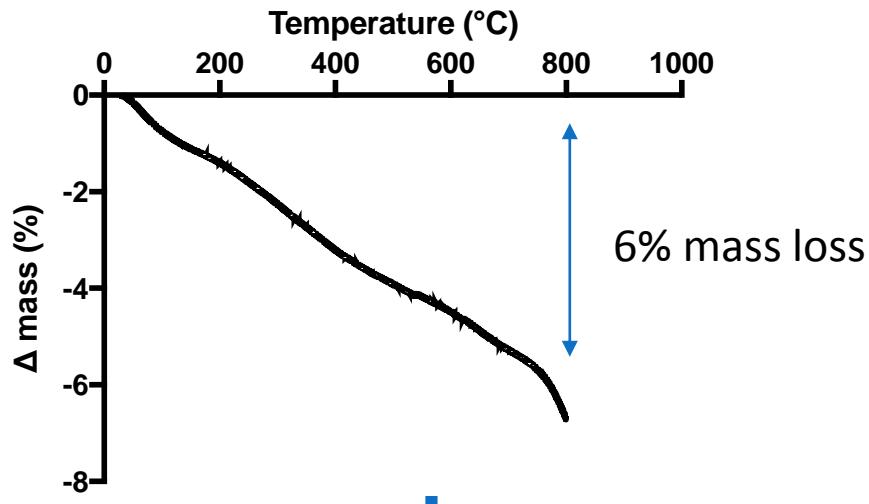
X-ray Fluorescence	
Chemical element	Composition (%)
Mg	1,6
Al	11,0
Si	13,1
S	6,9
K	2,1
Ca	3,5
Ti	1,3
Cr	1,8
Mn	0,2
Fe	29,1
Cu	16,7
Zn	7,7
Zr	1,0
Sn	1,1
Bi	1,6

Physicochemical characterization (1)

X-ray Fluorescence

Chemical element	Composition (%)
Mg	1,6
Al	11,0
Si	13,1
S	6,9
K	2,1
Ca	3,5
Ti	1,3
Cr	1,8
Mn	0,2
Fe	29,1
Cu	16,7
Zn	7,7
Zr	1,0
Sn	1,1
Bi	1,6

Thermogravimetric analysis



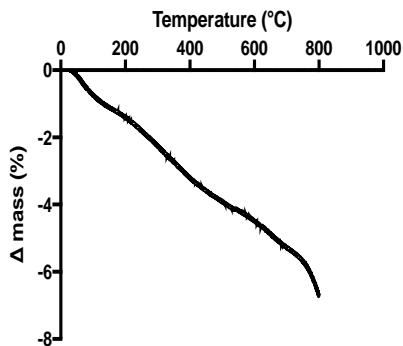
Organic : 6%
Inorganic : 94%

Physicochemical characterization (1)

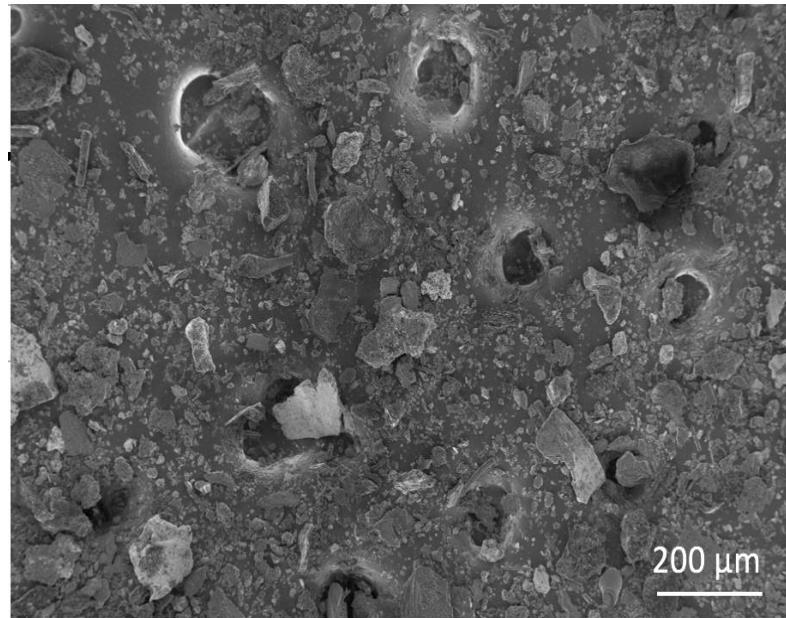
X-ray Fluorescence

Chemical element	Composition (%)
Mg	1,6
Al	11,0
Si	13,1
S	6,9
K	2,1
Ca	3,5
Ti	1,3
Cr	1,8
Mn	0,2
Fe	29,1
Cu	16,7
Zn	7,7
Zr	1,0
Sn	1,1
Bi	1,6

Thermogravimetric analysis



Scanning Electron Microscopy



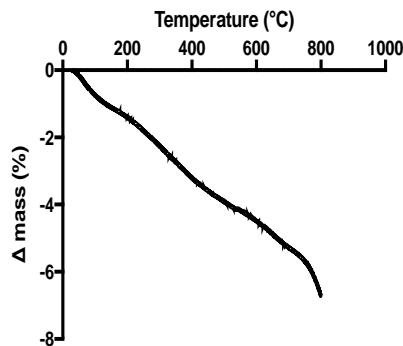
Organic : 6%
Inorganic : 94%

Physicochemical characterization (1)

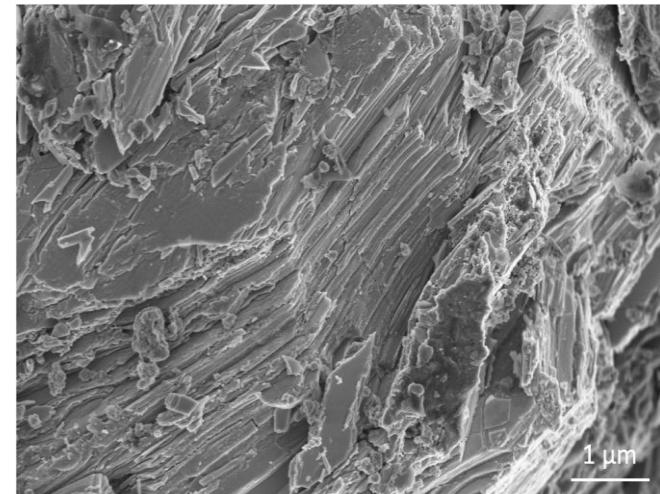
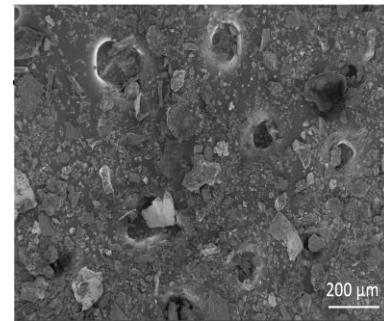
X-ray Fluorescence

Chemical element	Composition (%)
Mg	1,6
Al	11,0
Si	13,1
S	6,9
K	2,1
Ca	3,5
Ti	1,3
Cr	1,8
Mn	0,2
Fe	29,1
Cu	16,7
Zn	7,7
Zr	1,0
Sn	1,1
Bi	1,6

Thermogravimetric analysis



Scanning Electron Microscopy



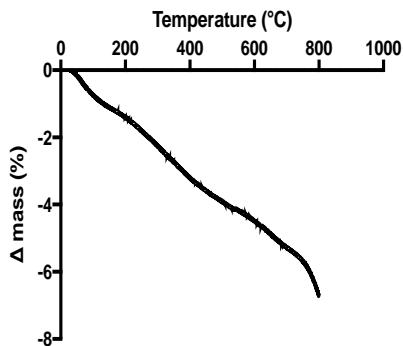
Organic : 6%
Inorganic : 94%

Physicochemical characterization (1)

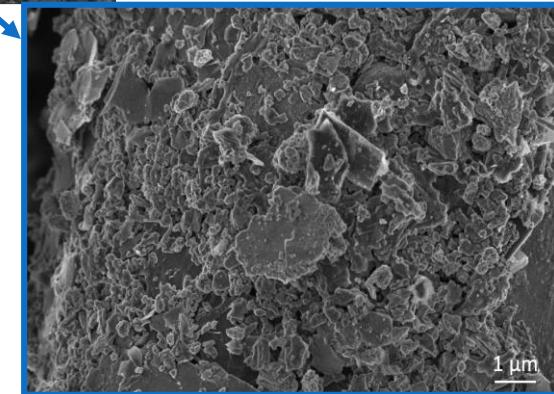
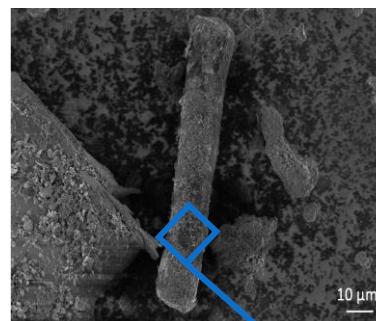
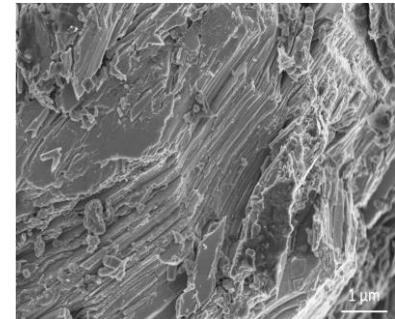
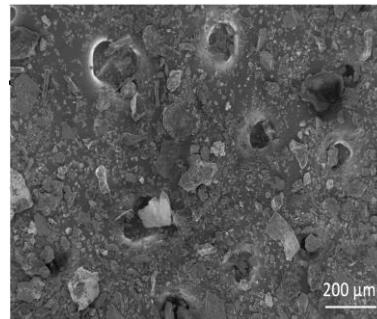
X-ray Fluorescence

Chemical element	Composition (%)
Mg	1,6
Al	11,0
Si	13,1
S	6,9
K	2,1
Ca	3,5
Ti	1,3
Cr	1,8
Mn	0,2
Fe	29,1
Cu	16,7
Zn	7,7
Zr	1,0
Sn	1,1
Bi	1,6

Thermogravimetric analysis



Scanning Electron Microscopy



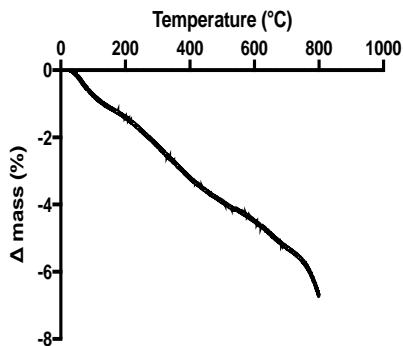
Organic : 6%
Inorganic : 94%

Physicochemical characterization (1)

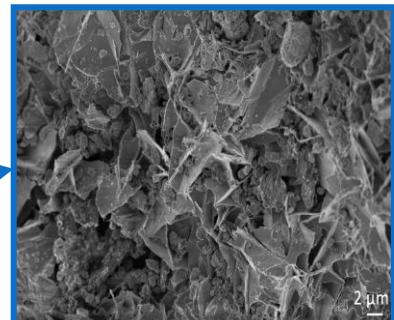
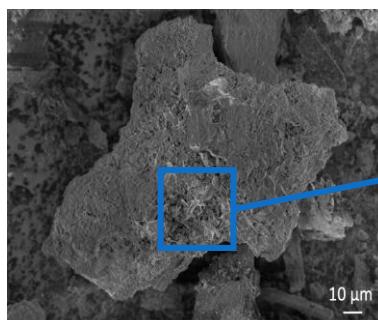
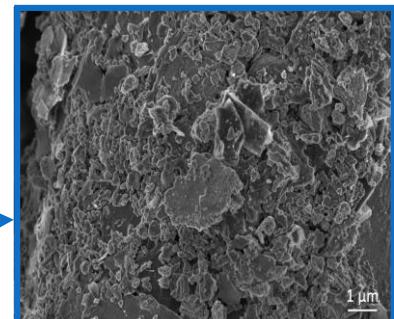
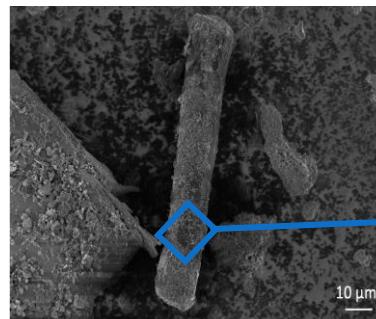
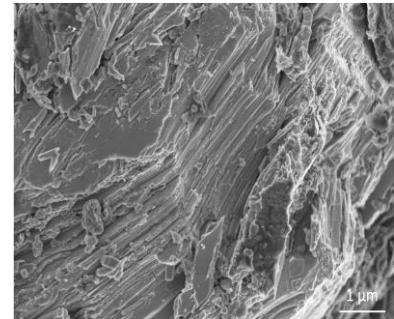
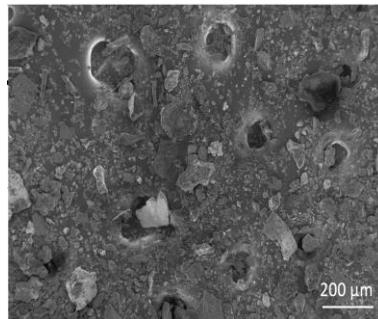
X-ray Fluorescence

Chemical element	Composition (%)
Mg	1,6
Al	11,0
Si	13,1
S	6,9
K	2,1
Ca	3,5
Ti	1,3
Cr	1,8
Mn	0,2
Fe	29,1
Cu	16,7
Zn	7,7
Zr	1,0
Sn	1,1
Bi	1,6

Thermogravimetric analysis



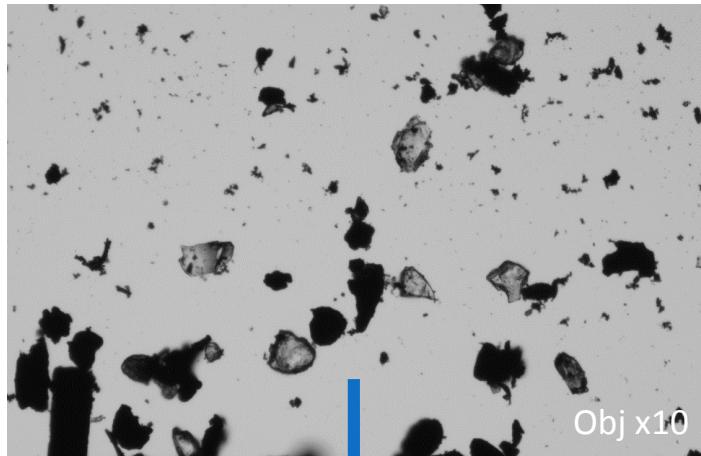
Scanning Electron Microscopy



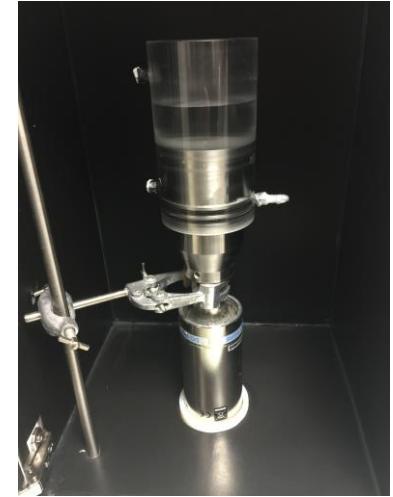
Organic : 6%
Inorganic : 94%

Physicochemical characterization (2)

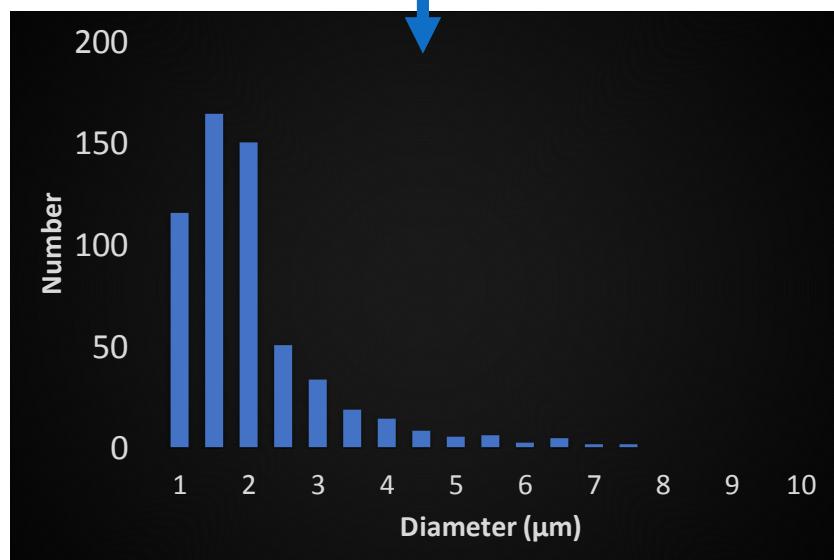
Optical microscopy



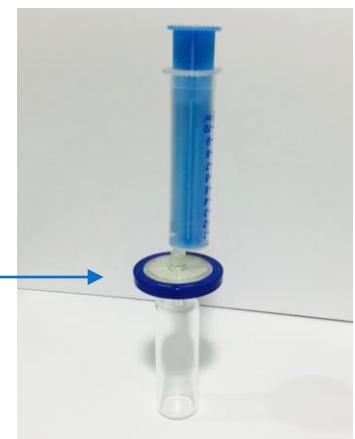
0 → 10 min



Sonication
+
Filtration
process



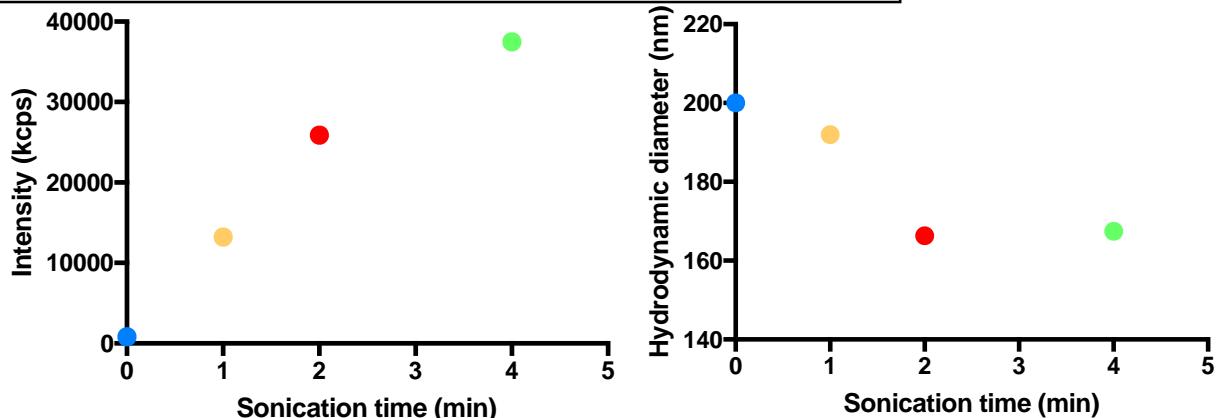
0,45 μm filter



Physicochemical characterization (3)

Characterization of brake wear powder suspensions

	Sonication time (min)	Unifiltered	Filtered
Z-average (d.nm)	0	12000	200
	1	1600	192
	2	1200	166
	4	621	167
Intensity (kcps)	0	$3,4 \times 10^5$	$8,1 \times 10^2$
	1	$6,7 \times 10^5$	$1,3 \times 10^5$
	2	$5,2 \times 10^5$	$2,5 \times 10^5$
	4	$1,8 \times 10^6$	$3,7 \times 10^5$



Sonication + filtration process :
Size decrease
Intensity increase



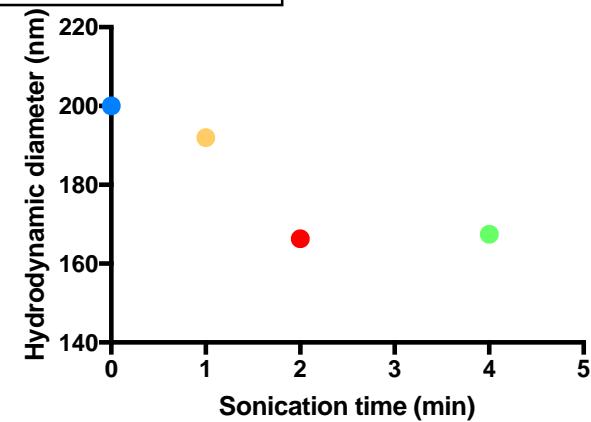
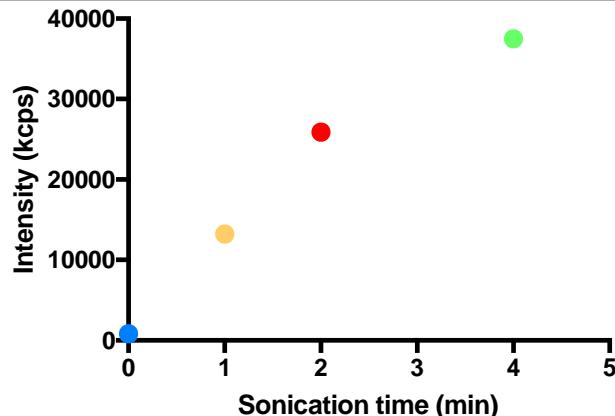
- ✓ Selection of nanometer size fraction
- ✓ Stability study

Physicochemical characterization (3)

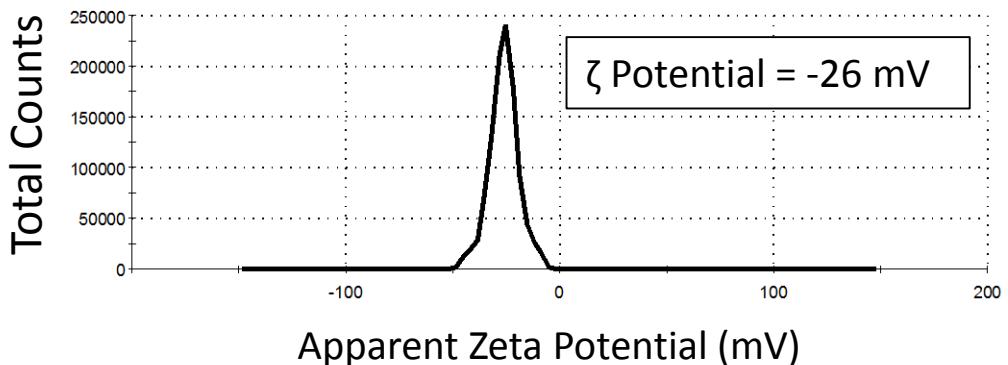
Characterization of brake wear powder suspensions

	Sonication time (min)	Unifiltered	Filtered
Z-average (d.nm)	0	12000	200
	1	1600	192
	2	1200	166
	4	621	167

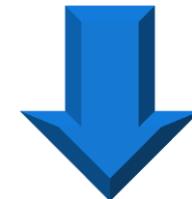
	Sonication time (min)	Intensity (kcps)
Intensity (kcps)	0	$3,4 \times 10^5$
	1	$6,7 \times 10^5$
	2	$5,2 \times 10^5$
	4	$1,8 \times 10^6$



Sonication + filtration process :
Size decrease
Intensity increase

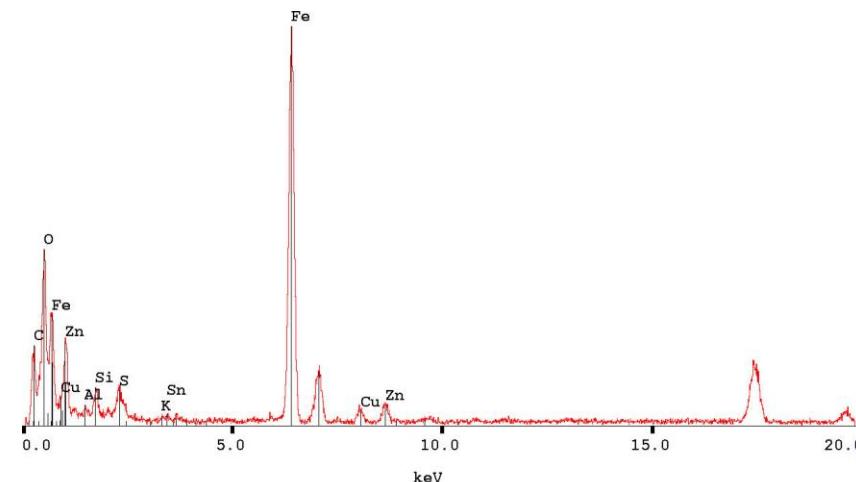
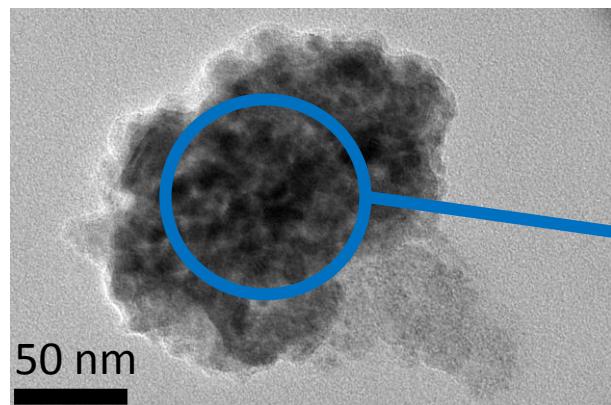
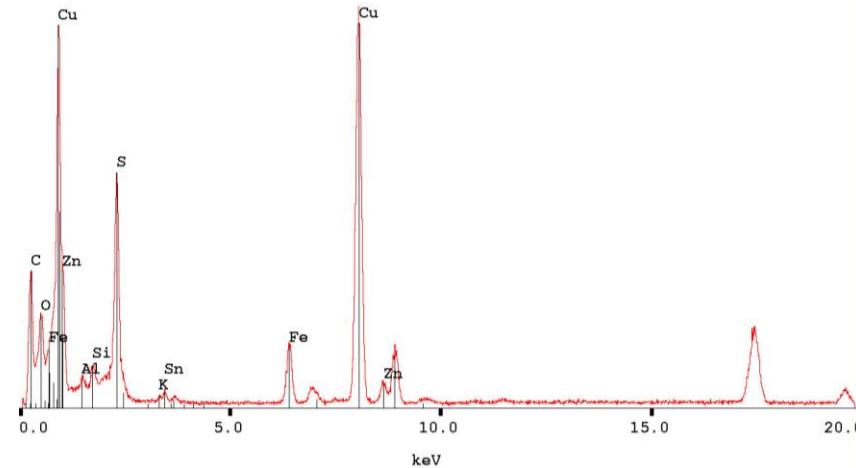
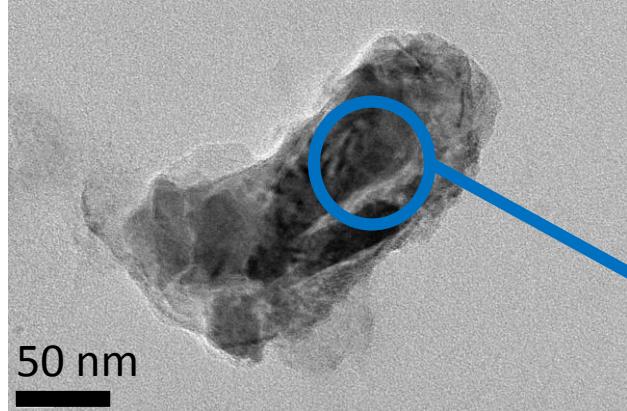


- ✓ Selection of nanometer size fraction
- ✓ Stability study



Physicochemical characterization (4)

Transmission electron microscopy coupled with energy dispersive X-ray spectroscopy of filtered suspensions

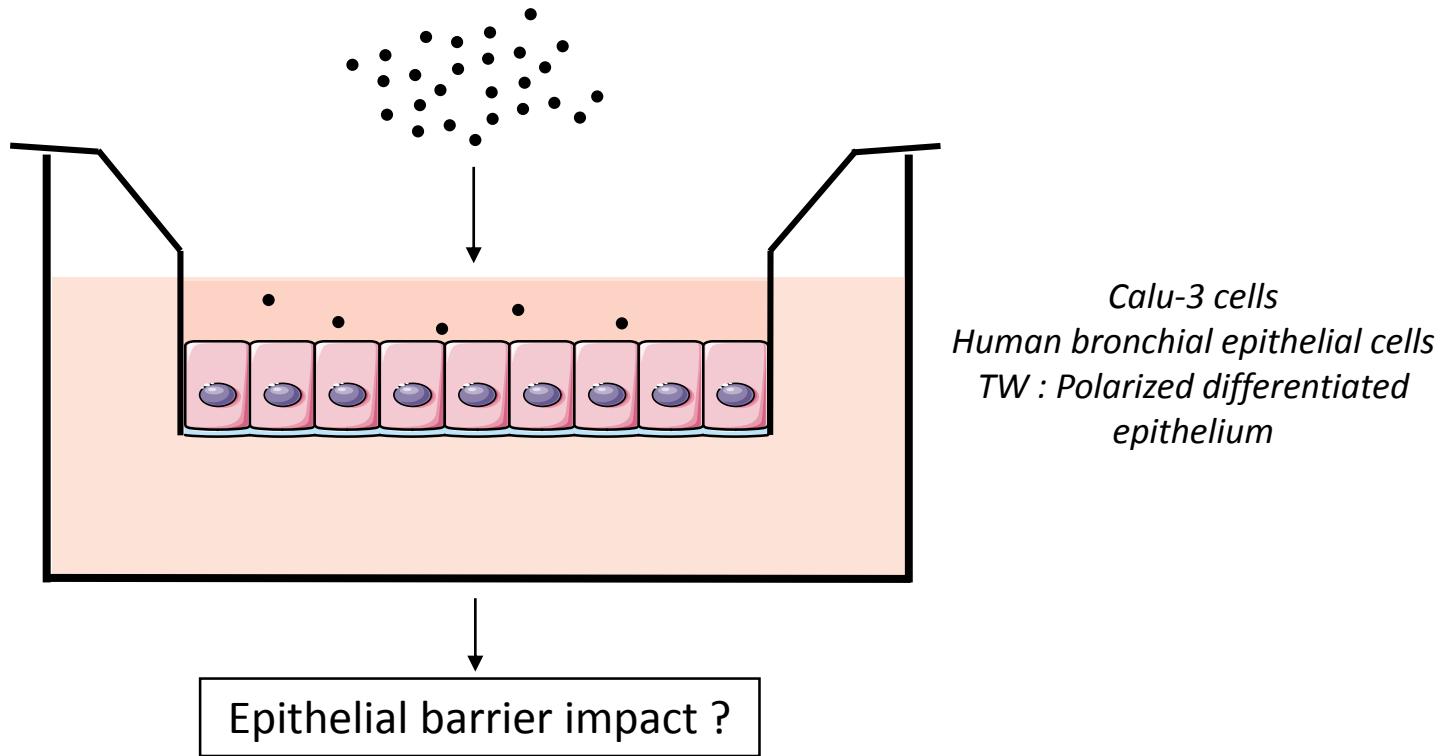


Toxicological study (1)

24h exposure to :

Brake wear powder suspensions

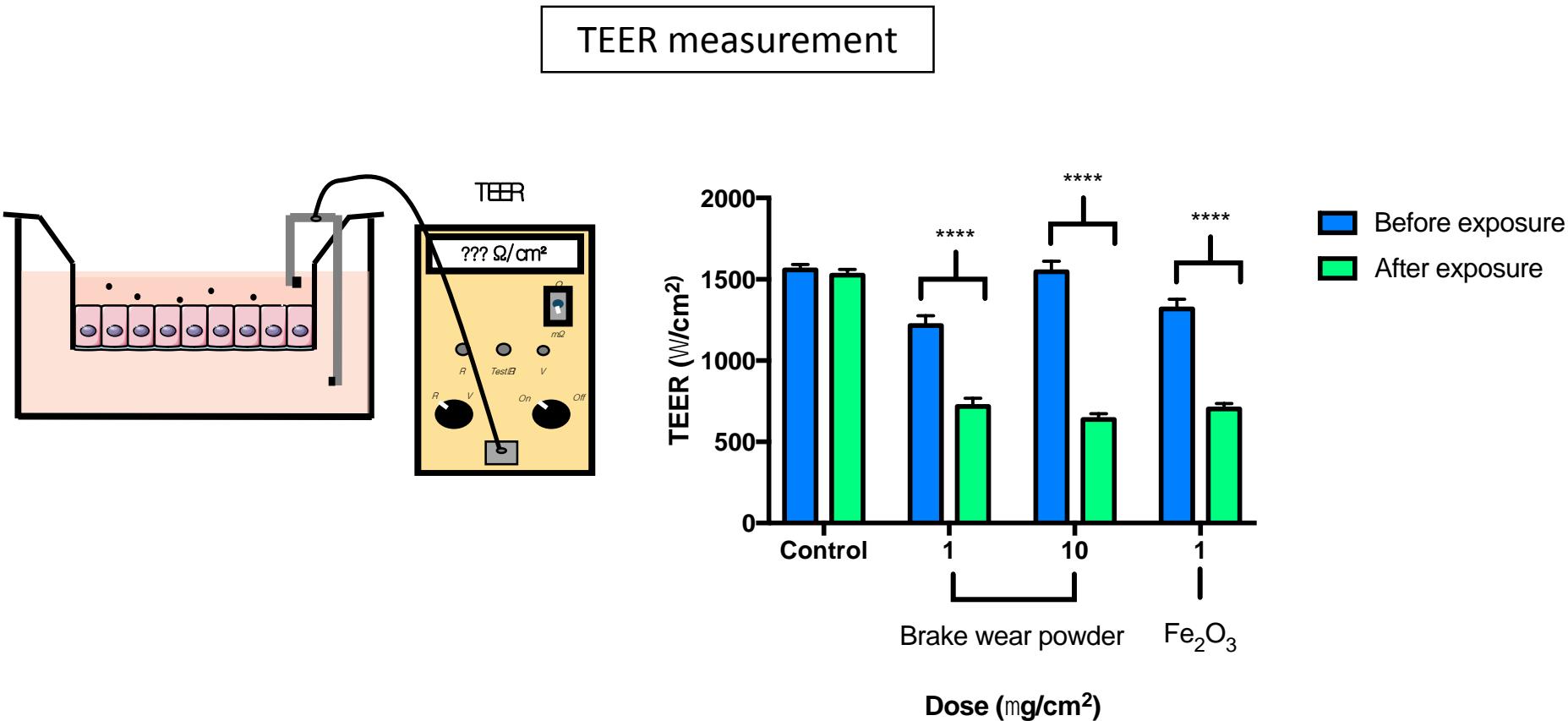
Benchmark = Fe_2O_3 nanoparticles (50 nm)



- **Barrier Integrity study** : Lucifer Yellow permeability assay, Transepithelial electric resistance measurement (TEER)
- **Cellular morphology** : Immunostaining (Actin, ZO-1, MUC5AC)
- **Proinflammatory response** : ELISA (IL-6, IL-8, TNF)

Toxicological study (2)

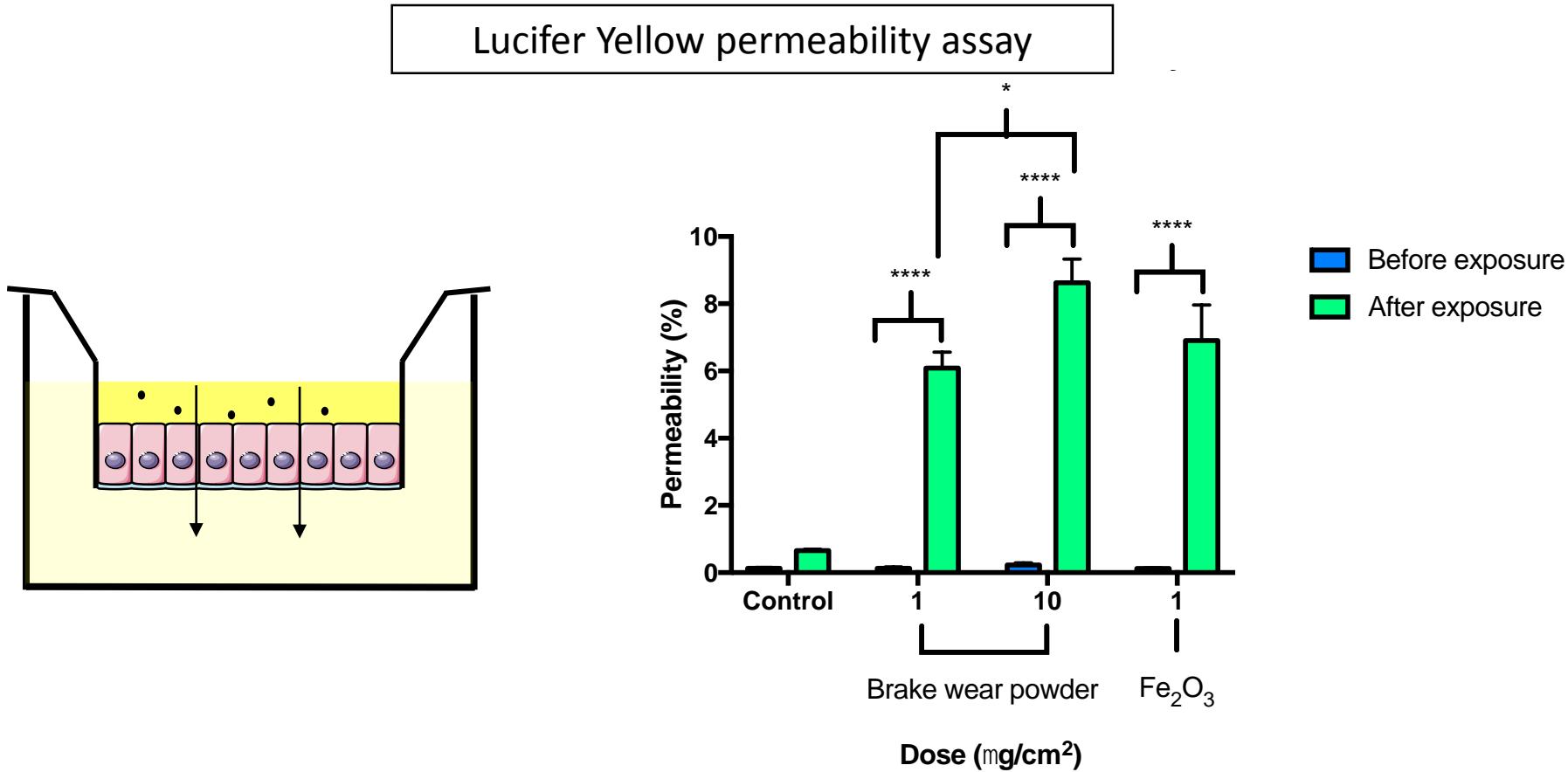
Epithelial barrier integrity assessment following 24h exposure to particles



Statistical analysis : ANOVA – Dunnett's post test, compared to vehicle, P<0,05 : *, P < 0,01 : **, P<0,001 : ***, P<0,0001 : ****

Toxicological study (2)

Epithelial barrier integrity assessment following 24h exposure to particles

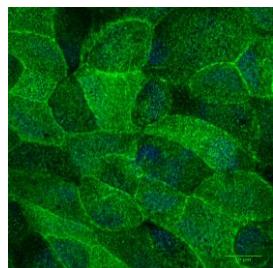


Statistical analysis : ANOVA – Dunnett's post test, compared to vehicle, $P < 0,05$: *, $P < 0,01$: **, $P < 0,001$: ***, $P < 0,0001$: ****

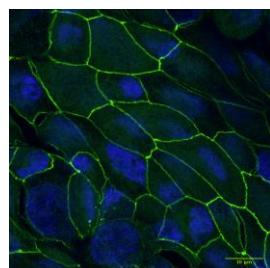
Toxicological study (3)

Vehicle

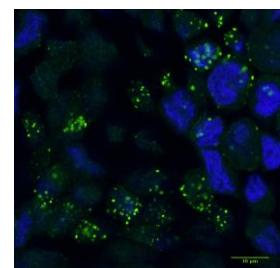
Actin



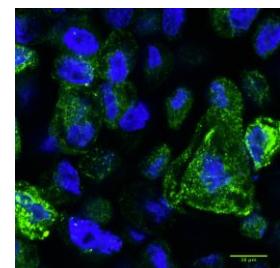
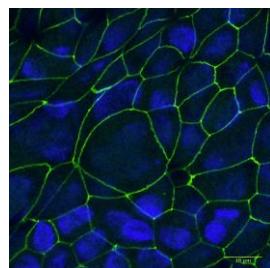
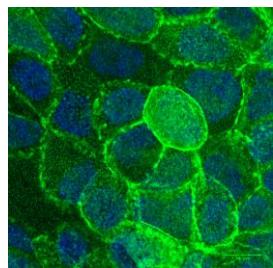
ZO-1



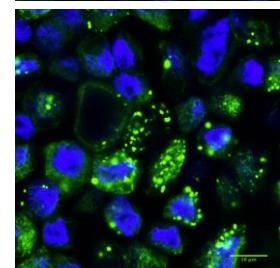
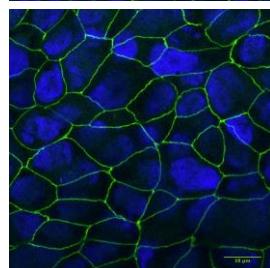
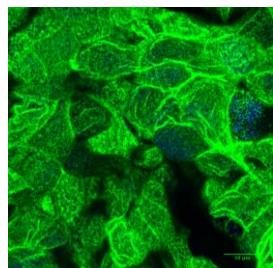
MUC5AC



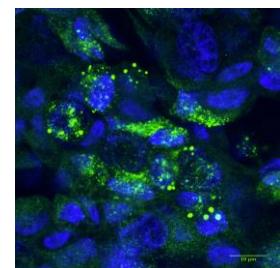
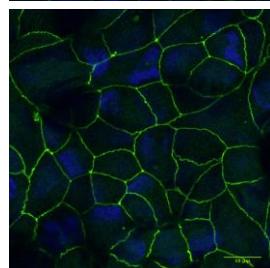
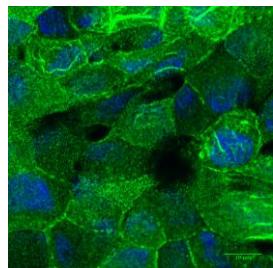
Brake wear powder
(1 µg/cm²)



Brake wear powder
(10 µg/cm²)



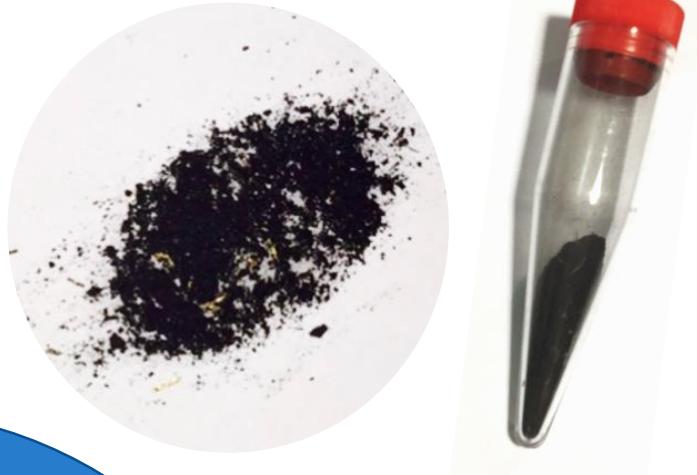
Fe₂O₃ nanoparticles
(1 µg/cm²)



Conclusions

Physico-chemical characterization

- Complex mixture
- Major components : Fe, Cu
- Micro- and Nano-sized fractions
- Different amorphous particles



Conclusions

Physico-chemical characterization

- Complex mixture
- Major components : Fe, Cu
- Micro- and Nano-sized fractions
- Different amorphous particles



Toxicological study

- Permeability ↑ = Barrier functions ↓
= tight junctions injury
- No pro-inflammatory response (data not shown)
- Mucus production ↑
- Response similar as benchmark particles

UMR CNRS 8251
Unité BFA, Laboratoire RMCX
Armelle BAEZA-SQUIBAN
Sonja BOLAND
Pascal ROUSSEL
Valentina SIRRI-ROUSSEL

Maï-Lan TRAN
Roseline NILA

Financial support

DIM Nano-K, Région Ile-de-France

UMR CNRS 7057
Laboratoire MSC
Jean-François BERRET
Evdokia OIKONOMOU
Fanny MOUSSEAU
Alexandra Sasha LANIECE
Victor BALDIM
Anh Thai LE PHUONG

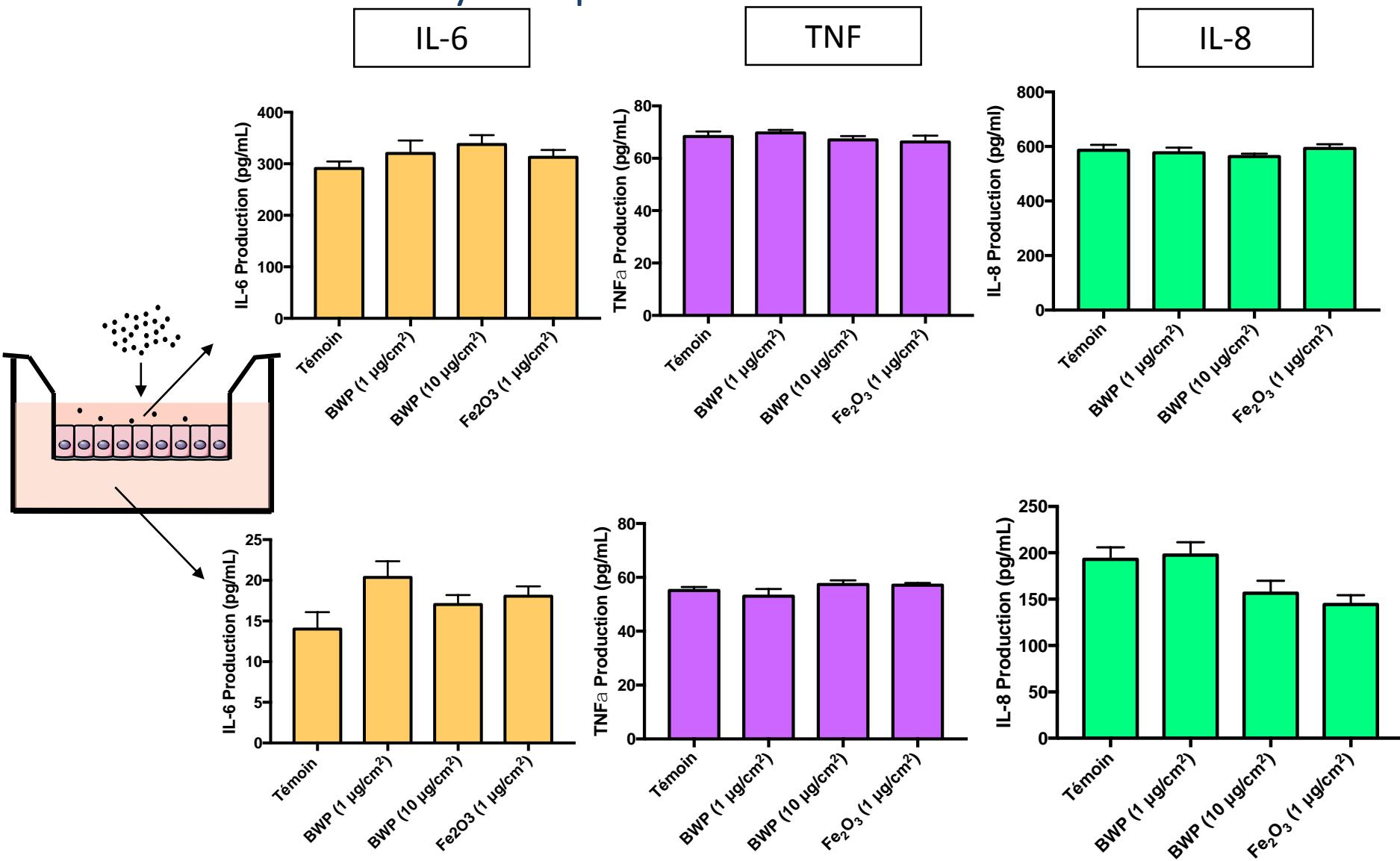
UMR CNRS 7086
Laboratoire ITODYS
Alexandre CHEVILLOT
Sophie NOWAK

Services de microscopie électronique

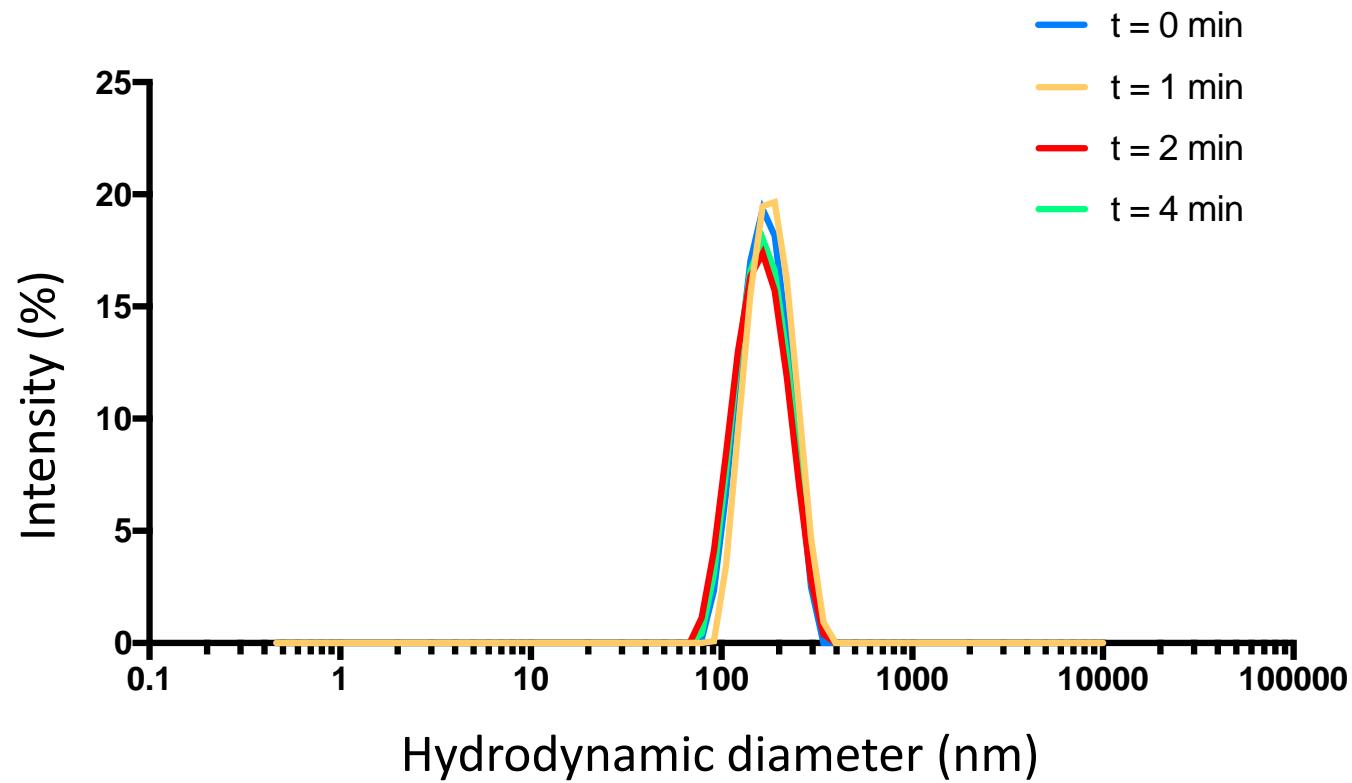
Sandra Casale (UPMC – TEM-EDX)

Ludovic MOUTON (UPD - SEM)

Pro-inflammatory response



Size distribution following sonication and filtration process



Stability study

